

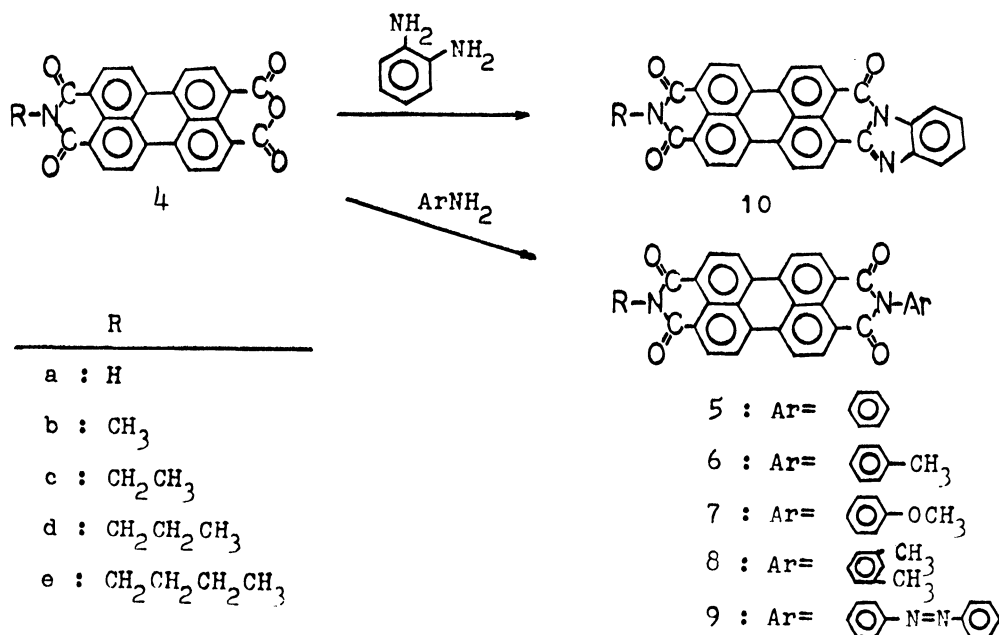
SYNTHESIS OF UNSYMMETRICAL PERYLENEBIS(DICARBOXIMIDE) DERIVATIVES

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Unsymmetrical perylenebis(dicarboximide) derivatives—N-alkyl-N'-aryl-3,4:9,10-perylenebis(dicarboximide) (5a-e, 6a-e, 7a-e, 8a-e, and 9a-e)—were prepared by the condensation of N-alkyl-3,4:9,10-perylenetetracarboxylic monoanhydride monoimide (4a-e) with arylamines (aniline, p-toluidine, p-anisidine, and 3,5-xylidine etc.).

In general, 3,4:9,10-perylenebis(dicarboximide) derivatives (PTCI) are synthesized by the condensation of 3,4:9,10-perylenetetracarboxylic dianhydride (PTCA)(1) with amines. PTCI have excellent fastness to light, heat, and solvent etc. Some of them are used as dyes or pigments. Since the working by M. Kordos¹⁾, Maki et al.^{2,3)}, and Bradley et al.⁴⁾, many kinds of N,N'-diaryl-PTCI(2) and some kinds of N,N'-dialkyl-PTCI(3) were obtained.

On the other hand, an unsymmetrical PTCI including different substituents on N and N' atoms have not been obtained. The preparation of the unsymmetrical PTCI was attempted by the formation of 3,4:9,10-perylenetetracarboxylic monoanhydride monoimide (PTCMI)(4) by the condensation of 1 with an alkylamine, followed by the condensation of 4 with an arylamine. However 4 obtained already was only 4a.⁵⁻⁹⁾ So in our



previous work the condensation of 1 with alkylamines had been studied and their corresponding N-alkyl-PTCMI(4a-e) were obtained.¹⁰⁾ In this work the condensation of 4 with arylamines(ArNH₂) were attempted and unsymmetrical PTCI(5-10) were prepared successfully.

The unsymmetrical PTCI(5-10) prepared by the condensation of PTCMI(4) with arylamines are listed in Table 1. Their structures were confirmed by elemental analyses, absorption spectra, and IR and MS spectra. These are given in Table 2.


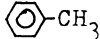
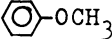
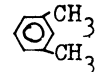
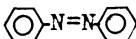
Table 1 Reaction conditions for The Condensation of N-Alkyl-PTCMI(4) with Arylamines(ArNH₂)

N-Alkyl-PTCMI(4)	Arylamine		Temp. (°C)	Time (h)	Product Yield			
	P	(g)			(g)	(%)		
4a	H	1.0	Aniline	39.6	181	6.5	5a	32
4b	CH ₃	1.0	"	39.5	"	5.0	5b	63
4c	C ₂ H ₅	1.0	"	81.5	183	5.0	5c	91
4d	n-C ₃ H ₇	0.5	"	81.5	"	6.0	5d	55
4e	n-C ₄ H ₉	1.0	"	81.5	"	7.0	5e	94
4a	H	1.0	p-Toluidine	20.6 ^{a)}	125	4.0	6a	85
4b	CH ₃	1.0	"	21.1 ^{a)}	"	5.0	6b	84
4c	C ₂ H ₅	0.5	"	8.2	"	5.0	6c	83
4d	n-C ₃ H ₇	0.5	"	8.2	"	5.0	6d	94
4e	n-C ₄ H ₉	1.0	"	15.4	"	3.0	6e	95
4a	H	1.0	p-Anisidine	15.0	187	5.0	7a	39
4b	CH ₃	1.0	"	15.2 ^{b)}	"	5.0	7b	83
4c	C ₂ H ₅	1.0	"	13.5	185	5.0	7c	94
4d	n-C ₃ H ₇	0.5	"	6.4	"	7.0	7d	77
4e	n-C ₄ H ₉	1.0	"	13.8	"	6.0	7e	95
4a	H	1.0	3,5-Xylidine	9.8 ^{b)}	190	7.0	8a	40
4b	CH ₃	1.0	"	9.8 ^{b)}	"	6.0	8b	99
4c	C ₂ H ₅	0.5	"	9.8 ^{b)}	"	7.0	8c	86
4d	n-C ₃ H ₇	0.5	"	9.8 ^{b)}	"	7.0	8d	55
4e	n-C ₄ H ₉	1.0	"	9.8 ^{b)}	"	7.0	8e	87
4a	H	1.0	p-Aminoazo- benzene	5.0 ^{c)}	233	6.0	9a	46
4b	CH ₃	1.0	"	5.0 ^{c)}	"	6.0	9b	63
4c	C ₂ H ₅	1.0	"	5.0 ^{d)}	283	5.0	9c	43
4d	n-C ₃ H ₇	1.0	"	5.0 ^{c)}	233	6.0	9d	47
4e	n-C ₄ H ₉	1.0	"	5.0 ^{d)}	288	6.0	9e	51
4a	H	1.0	o-Phenylene- diamine	3.7	180	4.0	10a	61
4b	CH ₃	1.0	"	3.5	"	4.0	10b	81

a) Added with 3.8 g of Acetic anhydride. b) Added with 16.4 g of DMF.

c) Added with 50 ml of Quinoline. d) Added with 15 g of Diphenylene oxide.

Table 2 Properties of Unsymmetrical PTCI

Unsymmetrical PTCI	R Ar		$\lambda_{\max}^b)$ (nm)	IF(KBr, cm^{-1})			MS Spectra (M^+ , m/e)	Color ^{c)}
				C=O (Imide)	C-H (Phenyl)			
5a	H		599	1701	1652	694	466	Red
5b	CH ₃	"	598	1692	1652	699	480	Dark reddish violet
5c	C ₂ H ₅	"	600	1698	1665	695	494	Red
5d	n-C ₃ H ₇	"	600	1701	1660	697	508	Red
5e	n-C ₄ H ₉	"	600	1703	1656	695	522	Red
6a	H		599	1700	1672	801 779	480	Dark violet
6b	CH ₃	"	599	1696	1655	792 771	494	Dark red
6c	C ₂ H ₅	"	601	1702	1656	804 770	508	Reddish violet
6d	n-C ₃ H ₇	"	602	1701	1660	789 767	522	Reddish violet
6e	n-C ₄ H ₉	"	602	1703	1659	798 785	536	Reddish violet
7a	H		598	1690	1660	800 779	496	Dark violet
7b	CH ₃	"	599	1694	1658	795 771	510	Dark red
7c	C ₂ H ₅	"	602	1699	1658	795 768	524	Dark violet
7d	n-C ₃ H ₇	"	602	1698	1657	794	538	Dark violet
7e	n-C ₄ H ₉	"	602	1698	1656	796 778	552	Dark violet
8a	H		600	1700	1661	799 682	494	Red
8b	CH ₃	"	600	1702	1661	798 680	508	Dark red
8c	C ₂ H ₅	"	602	1705	1666	805 686	522	Dark red
8d	n-C ₃ H ₇	"	602	1701	1666	799 681	536	Dark red
8e	n-C ₄ H ₉	"	602	1700	1664	796 680	550	Dark red
9a	H		602	1691	1653	768 687	570	Gray
9b	CH ₃	"	598	1703	1661	770 689	584	Dark reddish violet
9c	C ₂ H ₅	"	600	1700	1658	770 689	598	Dark violet
9d	n-C ₃ H ₇	"	599	1697	1657	768 685	612	Dark red
9e	n-C ₄ H ₉	"	600	1702	1660	770 638	625	Dark violet
10a	H		621	1705	1670	750	463	Dark violet
10b	CH ₃		626	1710	1670	755	477	Gray

a) C-O(Ether). b) Solvent: Conc. H₂SO₄. c) One g of finely powdered pigment, 2 g of Titanium dioxide, and 2 ml of Boiled Oil were mixed by a maller. The mixture was painted on a paper, then the Color was observed.

For example, 5a was prepared as follows. To a three necked flask were added 39.6 g of aniline and 1.0 g of 4a. The flask was heated at 181 °C for 6.5 h with stirring. To the cooled reaction mixture was added methanol, then the mixture was warmed, filtered, and washed with methanol to remove aniline. The residue was added into 1% hot potassium hydroxide solution and filtered to remove unchanged 4a. The precipitate was washed with methanol and dried. Reddish violet powder of 5a was obtained (32% of yield). 5b-e were prepared by the same treatment of 4b-e with aniline. 6a-e, 7a-e, 8a-e, and 9a-e were similarly prepared by the reaction of 4a-e with p-toluidine, p-anisidine, 3,5-xylidine, p-aminoazobenzene, respectively. These reaction conditions are listed in Table 1.

In comparison with the reaction with alkylamines, more drastic conditions were necessary for the reaction with arylamines. The basicities of arylamines (pka, aniline; 4.58¹¹⁾, p-toluidine; 5.08¹²⁾, p-anisidine; 5.34¹²⁾, o-phenylenediamine; 4.47¹³⁾) are lower than those of alkylamines, and the reaction with aniline under drastic conditions gave low yield of PTCI than with p-toluidine. Then effect in reactivity due to basicity of amine would be expected.

The properties of unsymmetrical PTCI and symmetrical PTCI were compared. IR spectra of unsymmetrical PTCI (5-9) showed the bands at 1690-1705 and 1650-1670 cm⁻¹ (C=O of imide). Symmetrical PTCI (2 and 3) showed the bands at 1695-1710 and 1655-1690 cm⁻¹ (C=O of imide). Absorption spectra of unsymmetrical PTCI (5-9) showed the λ_{max} at 598-602 nm. In symmetrical PTCI, 3a-e showed the λ_{max} at 596-597 nm and 2 showed λ_{max} at 600-606 nm. N-substituted alkyl or aryl group gave only a small influence on bands of C=O or λ_{max} . The colors of many unsymmetrical PTCI were reddish like those of symmetrical PTCI. Most of the unsymmetrical PTCI were not melted or decomposed by heating until 400 °C in air, and some of Unsymmetrical PTCI showed a very clear color, then the detailed test for pigments are now in progress.

References

- 1) M. Kordos, D.R.P., 276357, 276956 (1913).
- 2) T. Maki and H. Hashimoto, *Kogyo Kagaku Zasshi*, 54, 479 (1951); *ibid.*, 54, 544 (1951).
- 3) T. Maki and H. Hashimoto, *Bull. Chem. Soc. Jpn.*, 25, 411 (1952); *ibid.*, 27, 602 (1954).
- 4) W. Bradley, F. W. Pexton, *J. Chem. Soc.*, 4432 (1954).
- 5) M. P. Schmidt and W. Neugebauer, U.S.P., 1506545 (1924).
- 6) W. Neugebauer, D.R.P., 41217 (1925).
- 7) P. Koshits and I. S. Poulushenko, *J. Gem. Chem.*, 17, 1739 (1947).
- 8) T. Misono and Y. Nagao, *Yuki Gosei Kagaku Kyokai Shi*, 29, 317 (1971).
- 9) Y. Nagao and T. Misono, *Kogyo Kagaku Zasshi*, 71, 2500 (1971).
- 10) This will be published in *Nippon Kagaku Kaishi*.
- 11) N. F. Hall and M. R. Sprinkle, *J. Am. Chem. Soc.*, 64, 2588 (1942).
- 12) A. I. Biggs and R. A. Robinson, *J. Chem. Soc.*, 388 (1961).
- 13) R. Kuhn and A. Wasserman, *Helv. Chim. Acta*, 11, 1 (1928).

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